New Entry to  $\gamma$ -Butyrolactams by Free Radical Cyclization of N-Allyl- $\alpha$ -chloro- $\alpha$ -(methylthio)acetamides. Formal Total Synthesis of (±)-Pseudoheliotridane

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Tributyltin hydride-azobisisobutyronitrile induced radical cyclization of N-allyl- $\alpha$ -chloro- $\alpha$ -(methylthio)acetamides afforded  $\gamma$ -butyrolactams. This method was applied to the formal total synthesis of (±)-pseudoheliotridane.

Free radical cyclization is rapidly becoming an important synthetic method for cyclic compounds.  $^{1,2)}$  Although a number of radical species capable of ring closure with unsaturated bonds have been discovered so far, the use of sulfur-substituted radicals has received little attention.  $^{3)}$  Here, we wish to demonstrate the usefulness of the methylthio-substituted  $\alpha$ -carbamoyl radical ( $\underline{2}$ ) as an initiator for radical olefin cyclization which provides a new route to  $\gamma$ -butyrolactams.

In a typical experiment, a mixture of tributyltin hydride  $(\underline{n}-Bu_3SnH)$  (1.1 equiv.) and a catalytic quantity of azobisisobutyronitrile (AIBN) in benzene was injected over 30 min into a 0.06 M solution of the chloride  $(\underline{1a})^4$ ) in refluxing benzene, and refluxing was continued for 2 h. Evaporation of the solvent followed by flash chromatography on silica gel (benzene:ethyl acetate=4:3) gave 1,4-dimethyl-3-methylthiopyrrolidin-2(1H)-one (3a) in 68% yield as a mixture of two stereoisomers (trans:cis=69:31) (vide infra) [ $\delta$  (CDCl<sub>3</sub>, 300 MHz) 1.15 (d, J=6.7 Hz, CMe for cis), 1.22 (d, J=6.8 Hz, CMe for trans), 2.23 (s, SMe for trans), 2.26 (s, SMe for cis), 2.87 (3H, s, NMe)], along with the reduction product (4a) (16%). This result is in sharp contrast to a similar treatment of the chloride (5) which gave only a 24% yield of the cyclized product (7)<sup>5</sup>) together with the reduction product (8) (39%). It is generally accepted that the stabilized radicals are less reactive than the less stabilized radicals in the olefin cyclizations. This is,

<u>a</u>: R=Me, <u>b</u>: R=CH<sub>2</sub>Ph, <u>c</u>: R=Ph

Scheme 1.

however, in conflict with our results, since the radical (2) flanked by a pair of capto-dative substituents<sup>6)</sup> is expected to be more stable than the primary one  $(\underline{6})$ .

Assignment of <u>trans</u>-stereochemistry for the major isomer of the cyclization product ( $\underline{3a}$ ) was made on the basis of the thermodynamic consideration. Thus, treatment of the mixture of  $\underline{3a}$  with sodium ethoxide in refluxing ethanol resulted in an increase in the amount of the major isomer [ $\delta$  2.23 (s, SMe)] at the expense of the minor one [ $\delta$  2.26 (s, SMe)] (87:13 by  $^1\text{H-NMR}$ ). Earlier studies on the cyclizations of 1-substituted hex-5-enyl radicals have revealed that relatively stabilized radicals afford predominantly the <u>trans</u> products, while less stabilized radicals give <u>cis</u>-rich products. This is the case for the cyclization of  $\underline{2}$ .

The cyclizations of the chlorides  $(\underline{1b})$  and  $(\underline{1c})$  also gave the lactams  $(\underline{3b})$  (80%) and  $(\underline{3c})$  (90%) along with the reduction products  $(\underline{4b})$  (12%) and  $(\underline{4c})$  (8%), respectively. 9)

The 5-exo cyclization of the N-methallyl system  $(\underline{9})$  proceeded similarly, giving the pyrrolidinone  $(\underline{10})$  (68%) together with the reduction product  $(\underline{11})$  (15%). None of the 6-endo cyclization product was detected in the reaction mixture. In some cases, internal olefin substitution leads to enhanced endo cyclization for steric reason. The present example is a rare case of the exclusive formation of the exo-cyclization product.  $(\underline{11})$ 

MeN SMe 
$$n$$
-Bu<sub>3</sub>SnH MeN Me Me Me Me Me Me Me  $\frac{9}{11}$ 

Scheme 2.

Finally, we applied this method to the synthesis of a pyrrolizidine ring system. Thus, treatment of the chloride ( $\underline{14}$ ), which was prepared from  $\underline{L}$ -prolinol ( $\underline{12}$ ) as outlined in Scheme 3, with  $\underline{n}$ -Bu $_3$ SnH and AIBN gave the hexahydro-3 $\underline{H}$ -pyrrolizin-3-one ( $\underline{15}$ ) in 60% yield together with the reduction product ( $\underline{13}$ ) (24%). The lactam ( $\underline{15}$ ) was shown to be a mixture containing two or more diastereoisomers by  $^1$ H-NMR spectroscopy. Desulfurization of the compound ( $\underline{15}$ ) with Raney nickel afforded, in 80% yield, the  $1\alpha$ -methyl-lactam ( $\underline{16}$ ) [ $\delta$  1.16 (d, J=6.6 Hz, Me)], whose  $^1$ H-NMR spectrum (300 MHz) showed it to contain a trace amount ( $\langle 5 \rangle$ ) of the corresponding  $1\beta$ -methyl isomer ( $\underline{17}$ ) [ $\delta$  0.98 (d, J=7 Hz, Me)]. Chromatographic separation of these isomers and their reduction leading to ( $\pm$ )-pseudoheliotridane ( $\pm$ ) and ( $\pm$ )-heliotridane ( $\pm$ ), respectively, have been described in the literature.

Scheme 3. i, 
$$C1C0_2Et$$
, 4 M NaOH (91%); ii, DMSO, (COC1)<sub>2</sub>,  $Et_3N$ ,  $CH_2Cl_2$ ,  $-60$  °C (90%); iii,  $Ph_3PMe$  Br<sup>-</sup>,  $NaCH_2S(0)Me$ , DMSO (83%); iv, KOH,  $NH_2NH_2 \cdot H_2O$ , ( $CH_2OH$ )<sub>2</sub>, reflux; v, MeSCH<sub>2</sub>COCl,  $Et_3N$ ,  $Et_2O$  (iv and v, total 60%); vi, NCS,  $CHCl_3$  (quant.); vii,  $n-Bu_3SnH$ , AIBN,  $C_6H_6$ , reflux; viii, Raney Ni (W-2),  $EtOH$ , reflux.

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- 4) The chlorides ( $\underline{1}$ ) were prepared by acylation of the corresponding allylamines with  $\alpha$ -(methylthio)acetyl chloride and Et $_3$ N in diethyl ether followed by treatment of the resultant amides ( $\underline{4}$ ) with  $\underline{\text{N}}$ -chlorosuccinimide (NCS) in CCl $_4$ .
- 5) IR (CCl<sub>4</sub>):  $1690 \text{ cm}^{-1}$ ,  $^{1}\text{H-NMR}$  (CDCl<sub>3</sub>, 300 MHz)  $\delta$ : 1.13 (3H, d, J=6.7 Hz, CMe), 2.01 (1H, dd, J=16.0, 6.4 Hz, one of COCH<sub>2</sub>), 2.35-2.54 (1H, m, CHMe), 2.55 (1H, dd, J=16.0, 8.6 Hz, one of COCH<sub>2</sub>), 2.83 (3H, s, NMe), 2.96 (1H, dd, J=9.6, 5.9 Hz, one of NCH<sub>2</sub>), 3.49 (1H, dd, J=9.6, 7.7 Hz, one of NCH<sub>2</sub>). This compound was identical with that obtained by desulfurization (Raney Ni) of 3a.
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